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06ME32A

Third Semester B.E. Degree Examination, June 2012
Material Science and Metallurgy

Time: 3 hrs.

Max. Marks: 100

**Note: Answer FIVE full questions, selecting
at least TWO questions from each part.**

PART – A

- 1 a. Define atomic packing factor. Determine atomic packing factor of HCP crystal structure. (10 Marks)
- b. A 0.2% C steel component is to be carburised at 920°C. Calculate the time required to increase the carbon content to 0.4% at 0.5mm below the surface. Assume that carbon content at the surface is 0.9%. Given $D_{920^{\circ}\text{C}} = 1.28 \times 10^{-11} \text{ m}^2/\text{sec}$. (10 Marks)
- Error function values
- | Z | erf(z) |
|------|--------|
| 0.75 | 0.7112 |
| 0.80 | 0.7421 |
- 2 a. Draw the stress-strain curve for M.S. and label various points and explain them. (10 Marks)
- b. Explain the plastic deformation of metals and mechanisms that contribute to it. (10 Marks)
- 3 a. Explain with neat sketch the different stages of creep formation. (10 Marks)
- b. What is fatigue? Draw SN curves for
 i) Materials that display fatigue limit ii) Materials that do not display fatigue limit. (10 Marks)
- 4 a. What is solid solution? With suitable examples, explain the different types of solid solutions. (06 Marks)
- b. Describe the construction of phase diagram by thermal analysis. (06 Marks)
- c. Explain how the interpretation of phase diagrams is done. (08 Marks)

PART – B

- 5 a. Explain three types of invariant reactions occurring in iron carbon diagram, with Gibb's phase rule. (08 Marks)
- b. Explain how TTT diagrams are constructed. (06 Marks)
- c. Explain the microstructure of steel at 0.83 and 1.2% C. (06 Marks)
- 6 a. Define hardenability. Explain the Jominey end quench test, with related figures. (10 Marks)
- b. Explain the Austempering and martempering, with figure. (10 Marks)
- 7 a. Discuss AISI-SAE designation of steels, with examples. (08 Marks)
- b. Show schematically, the microstructures of cast iron, gray cast iron, white iron, malleable iron, ductile iron and compacted graphite iron. (12 Marks)
- 8 a. Explain general methods of corrosion prevention. (12 Marks)
- b. Write short notes on:
 i) Stress corrosion cracking
 ii) Cavitation damage. (08 Marks)

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Important Note : 1. On completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages.
 2. Any revealing of identification, appeal to evaluator and /or equations written eg, 42+8 = 50, will be treated as malpractice.